A Ready Synthesis of (\pm) -D-Homo-oestrone

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Summary Bis-annelation of the Wieland-Miescher ketone via 6-vinyl-α-picoline provides a convenient synthesis of (\pm) -D-homo-oestrone.

The bis-annelation of cyclohexanone has been achieved via base-induced hydrolytic cyclization of a dihydropyridine.¹ Herein we show an important application of bis-annelation in a ready synthesis of (\pm) -D-homo-oestrone.

Alkylation of the acetal-enone (I)² with 6-vinyl-α-picoline (II) [1 equiv. each of (I), (II), and potassium t-pentoxide in t-pentyl alcohol under reflux for 24 h] followed by acidic deacetalization gave (III)+ (70%). Reduction (NaBH₄) of the latter gave (IV), m.p. 102-105°. Catalytic hydro-(H₂-Pd-C-Et₃N-AcOEt) and acetalization afforded (V),† m.p. 128-130° [28% from (III)], which was smoothly transformed into (90%) the seco-hydroxyenedione (VI) \dagger by reduction [Na-NH₃ (1·2 equiv.)-EtOH (1·0 equiv.)], hydrolytic cyclization [NaOH-aqueous EtOH (3 equiv.); room temp.; 3 h], and deacetalization (aqueous HCl, 45°). Jones oxidation of (VI) gave enetrione (VIII),† m.p. 100-102°, which upon cyclodehydration (NaOEt-EtOH; reflux) gave (VIII), m.p. 162-164° (45%). The latter has previously been isomerized to (\pm)-D-homooestrone.3

Within the limits of detectability, (IX) was not a product of the hydrolytic cyclization of the dihydro-derivative of (V). It was of interest to determine the course of basecatalysed aldolization of the bona fide diketone (X), and so (V) was reduced as before but then treated with aqueous NaOH for 1 min. Compound (X)† was obtained (ca. 40%). Treatment of (X) with aqueous-ethanolic NaOH (3 equiv.) cleanly gave (XI) with no detectable (XII). Furthermore, attempted isomerization of (XI) to (XII)4 (hydration ≥ reverse aldol ≥ aldol ≥ dehydration) by heating the former in aqueous ethanolic alkali gave only recovered starting material. Finally, aldolization of (X) under reflux conditions again gave only (XI).

The overall conversion $(V) \rightarrow (VI)$ is thus consistent with the intermediacy of (X). The usual propensity for 7substituted heptane-2,6-diones to provide 2,3-disubstituted cyclohexenones4,5 is dramatically reversed in the case of (X), possibly as a consequence of a hitherto unknown branching effect which raises the energy of the pathway leading to (XII). The generality of these findings is discussed further in the following communication.6 Application of this new route to 19-norsteroids is under investigation.

$$(I) \times = -0[CH_2]_2O^-; R = H$$

$$(II) \times = -0[CH_2]_2O^-; R = H$$

$$(III) \times = 0; R = A_1 + A_2$$

$$(III) \times = 0 + A_2CH_2^-$$

$$(I$$

 $(X)X = \beta - OH$, $\alpha - H$; $Y = -O[CH_2]_2O_1R = Ac[CH_2]_3C(O)$ -(XI) $X = \beta - OH$, $\alpha - H$; $Y = -O[CH_2]_2O-$ (XII) Y = -0[CH₂]₂O-

This research was supported by Public Health Service grants. Mass spectra were measured by Mr. Richard Montgomery.

(Received, 30th December 1971; Com. 2207.)

[†] The structures of all new compounds are based on their i.r., n.m.r. and mass spectra. All enones were also characterized by their u.v. spectra. Combustion analyses within 0.3% of theory were obtained for all crystalline products.

S. Danishefsky and R. Cavanaugh, J. Amer. Chem. Soc., 1968, 90, 520.
 For the best preparation see: J. E. McMurry, J. Amer. Chem. Soc., 1968, 90, 6821.
 S. H. Douglas, J. M. Graves, D. Hartley, G. A. Hughes, B. J. McLoughlin, J. Siddall, and H. Smith, J. Chem. Soc., 1963, 5072.

⁴ Cf. R. N. Lacey, J. Chem. Soc., 1960, 1639.
⁵ G. Stork and R. Borch, J. Amer. Chem. Soc., 1964, 86, 935.
⁶ S. Danishefsky, A. Nagel, and D. Peterson, following communication.